

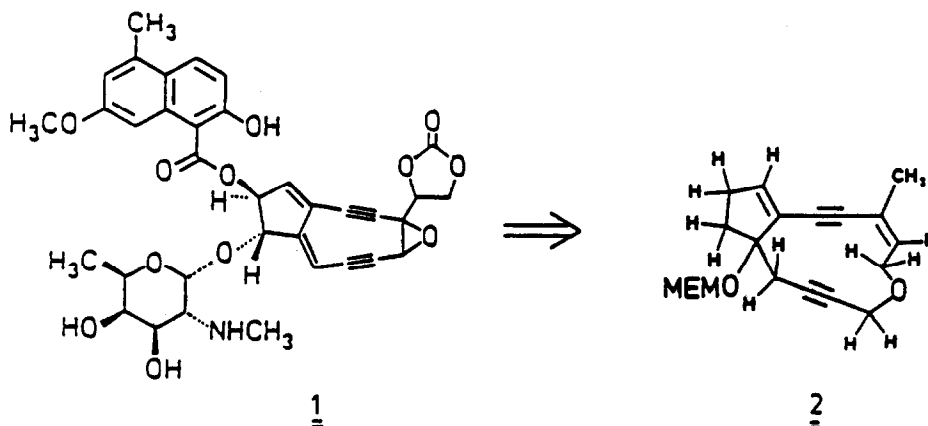
**SYNTHESIS OF A BICYCLIC OXACYCLOALKENEDIYNE SYSTEM
RELATED TO NEOCARZINOSTATIN CHROMOPHORE A**

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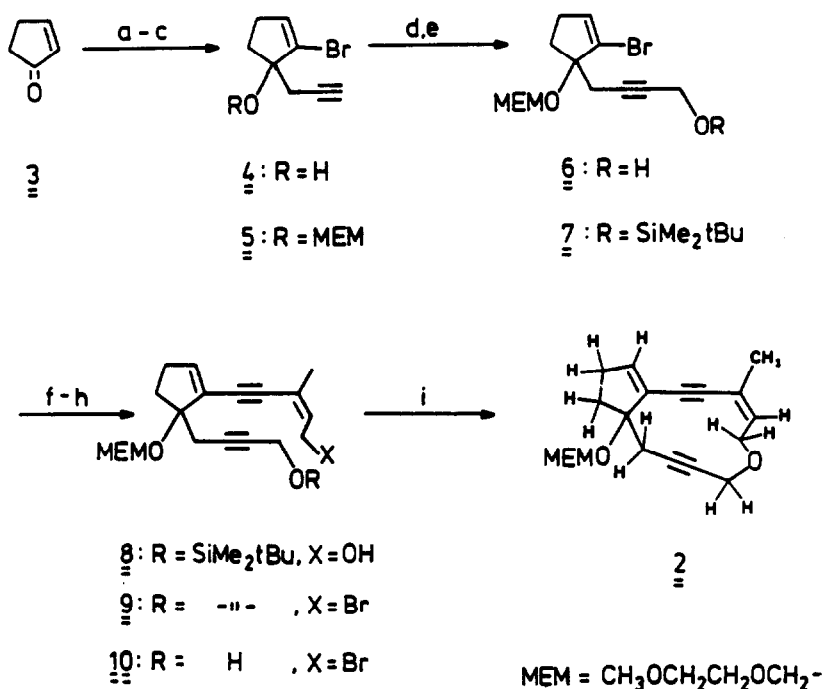
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Abstract: The synthesis of a stable bicyclic 12-membered oxacycloalkenedi-
diyne 2 related to the neocarzinostatin chromophore A 1 is described.

Neocarzinostatin chromophore ¹ (NCS-Chrom) is a member of a novel class of powerful antitumor antibiotics with a highly unusual cycloalkenedi-
diyne structure. Other examples are esperamicin,^{2a} calicheamicin ^{2b} and dynemicin A.³ Responsible for the biological activity of NCS-Chrom is a unprecedented, highly strained epoxybicyclo[7.3.0]dodecadienedi-
diyne system. The structure of NCS-Chrom was elucidated by Edo and coworkers.⁴ The proposed mechanism for DNA damage involves activation of the chromophore by a thiol function. Generation of a very reactive biradical which abstracts a hydrogen from C-5' of the DNA backbone induces strand scission upon aerobic incubation.⁵



Not only due to the extraordinary potency toward diseased cells but also because of the unexpected structure of this natural product, it has been a great challenge to find a useful synthetic approach to the bicyclic core of NCS-Chrom. The first synthesis of the parent bicyclic subunit of NCS-Chrom has been published by Wender et al.⁶ and recently Hirama and co-workers have synthesized stable 10-membered analogues.⁷ Our synthetic effort is concentrated on the synthesis of a functionalized core of NCS-Chrom and related model systems. Here we describe the synthesis of a stable bicyclic oxaalkenediyne system, starting from cyclopentenone 3.



a) Br_2 , Et_3N , CH_2Cl_2 , $0^\circ\text{C} \rightarrow \text{RT}$; b) $\text{HCCCH}_2\text{MgBr}$, Et_2O , RT ; c) MEMCl , $(\text{iPr})_2\text{NET}$, CH_2Cl_2 , 77%; d) LDA , -78°C ; $[\text{CH}_2\text{O}]_n$, $-78^\circ\text{C} \rightarrow \text{RT}$, 75%; e) TBDMSiCl , imidazole, DMF , 96%; f) $\text{HC}\equiv\text{C}(\text{CH}_3)\text{C}=\text{CCH}_2\text{OH}$, $\text{Pd}(\text{PPh}_3)_4$, CuI , PrNH_2 , 92%; g) SMe_2 , NBS , CH_2Cl_2 ; 0°C , 61%; h) Bu_4NF , THF , 0°C , 54%; i) 50% aq NaOH , TEBA-Cl , Et_2O , RT , 50%.

Bromination and dehydrobromination of the commercially available cyclopentenone 3 gave the bromocyclopentenone 8 which was exposed to propargylmagnesium bromide to give the acetylene 4.⁶ Protection of the tertiary alcohol with methoxyethyloxymethyl chloride ⁹ (MEMCl) provided 5 in 77% yield. Transmetalation with lithium diisopropylamide (LDA) and condensation with paraformaldehyde afforded the alcohol 6 in 75% yield followed by silylation with tert.butyltrimethylsilyl chloride which was achieved in 96% yield. Palladium catalyzed coupling of the vinyl bromide 7 with 1-hydroxy-3-methyl-2(Z)penten-4-yne gave the bisacetylenic alcohol 8 in excellent yield (92%). Cocatalysis with CuI is essential for this reaction.¹⁰ Treatment of 8 with dimethyl sulfide, N-bromosuccinimide in methylene chloride ¹¹ afforded the bromide 9 in 61% yield. Desilylation with tetrabutylammonium fluoride yielded the alcohol 10 (54%) as a potential precursor for the cyclisation. Slow addition of 10 to a vigorously stirred solution of aqueous NaOH (50%), diethyl ether and benzyltriethylammonium chloride as phase transfer catalyst gave 2 as a pale yellow oil in 50% yield.

The cycloalkenediyne 2 is characterized by spectroscopic data.¹² Most revealing was the detailed ¹H-NMR- and H,H-COSY-NMR-spectrum. It is interesting that one of the allylic protons appears at 4.05 ppm and the other at 4.57 ppm. This high field shift is in accord with the structure of 2 because one of the allylic protons is located in the anisotropic shielding cone of the two triple bonds.

Hitherto a novel route to the 9-membered core of the NCS-Chrom via a [2,3] sigmatropic Wittig-ring contraction has failed.

Further studies toward the synthesis of neocarzinostatin chromophore aglycone and model systems are in progress in our laboratories.

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- 12) All compounds were characterized by $^1\text{H-NMR}$, $^{13}\text{C-NMR}$, CI-MS, IR spectroscopy and elemental analysis.

2: $^1\text{H-NMR}$ (CDCl_3 , 250 MHz):

δ = 1.72-1.82 (m, 1H, CH_2), 1.8 (d, $^4J = 1.59$ Hz, 3H, CH_3), 2.20-2.53 (m, 3H, CH_2), 2.42 (dt, $J_{\text{AB}} = 17.07$ Hz, $J_{\text{AX}} = J_{\text{BX}} = 1.79$ Hz, 1H, $\text{CH-C}\equiv\text{C}$), 2.84 (dt, 2H, $J_{\text{AB}} = 17.07$ Hz, $J_{\text{AZ}} = J_{\text{BZ}} = 2.18$ Hz, 1H, $\text{CH-C}\equiv\text{C}$), 3.32 (s, 3H, OCH_3), 3.50 (t, 2H, $\text{OCH}_2\text{CH}_2\text{OCH}_3$), 3.60-3.90 (m, 2H, $\text{OCH}_2\text{CH}_2\text{OCH}_3$), 4.01, 4.11 (mc, $J_{\text{XZ}} = 15.88$ Hz, $J_{\text{AX}} = J_{\text{BX}} = 1.79$ Hz, $J_{\text{AZ}} = J_{\text{BZ}} = 2.18$ Hz, 2H, $\text{C}\equiv\text{CCH}_2\text{O}$), 4.05 (m, 1H, $\text{C}\equiv\text{CCHO}$), 4.57 (dd, $^2J = 10.72$ Hz, $^3J = 8.54$ Hz, 1H, $\text{C}\equiv\text{CCHO}$), 4.84 (AB, 2H, OCH_2O), 5.75 (m, $^3J = 8.54$ Hz, $^4J = 8.59$ Hz, 1H, $\text{H}_3\text{CC}=\text{CH}$), 6.26 (t, $^3J = 2.7$ Hz, 1H, olefin.H) ppm.

$^{13}\text{C-NMR}$ (CDCl_3 , 62.896 MHz):

δ = 22.59 (1C, CH_3), 30.05, 30.79 (each 1C, CH_2), 36.17 (1C, $\text{CH}_2\text{C}\equiv\text{C}$), 56.04 (1C, $\text{C}\equiv\text{CCH}_2\text{O}$), 58.97 (1C, OCH_3), 64.63 (1C, $\text{C}\equiv\text{CCH}_2\text{O}$), 67.36, 71.78 (each 1C, $\text{OCH}_2\text{CH}_2\text{O}$), 78.54, 84.24, 89.06, 89.79, (each 1C, $\text{C}\equiv\text{C}$), 91.45 (1C, OCH_2O), 91.50 (1C, C_q), 126.96 (1C, $\text{C}\equiv\text{C-C}$), 127.89 (1C, $\text{HC}=\text{CR}_2$), 131.48 (1C, $\text{C}\equiv\text{C-C}$), 142.06 (1C, $\text{HC}=\text{CR}_2$) ppm.

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